City College

B.Sc Semester 1, Internal Assessment(nline), under CU 2020-21

CHEMISTRY- HONOURS Paper-CC-1-1 (Organic chemistry) **Full Marks-10** Attempt all questions

- 1. Which compound does **not** have a conjugated system?

 - (A) $CH_2=CH-C\equiv C-CH_3$ (B) $CH_3-CH=C=CH-CH_3$

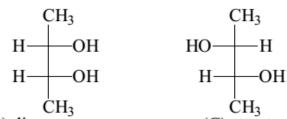
 - (C) CH₃-CH=CH-OCH₃ (D) CH₃-CH=CH-CN
- 2. Consider the S_N1 reaction of tert-butyl chloride with iodide ion:

$$(CH_3)_3C-CI + I^{\odot} \longrightarrow (CH_3)_3C-I + CI^{\odot}$$

If the concentration of iodide ion is doubled, the rate of forming tert-butyl iodide will: (hint: consider mechanism, i.e. how is the product formed?)

- (A) Double.
- (B) Increase 4 times.
- (C) Remain the same.

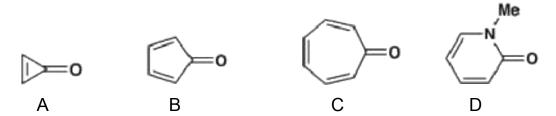
- (D) Decrease.
- (E) None of the above.
- 3. The relationship between the following two structures is:



- (A) enantiomers
- (B) diastereomers
- (C) structural isomers

- (D) identical
- (E) none of the above

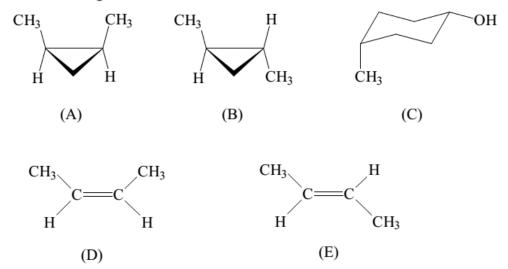
4. Which compound is not aromatic?



- 5. Which of the following alkyl halides would undergo S_N2 reaction most rapidly?
 - (A) CH₃CH₂-Br
- (B) CH₃CH₂- Cl
- (C) CH₃CH₂-I

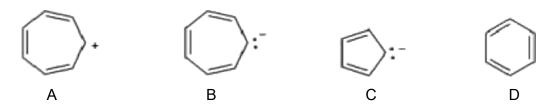
(D) CH₃CH₂-F

- (E) they react at the same rate
- 6. Which of the following molecules is chiral?



- 7. The relationship between (A) and (B) structures in the previous question is:
 - A) identical
 - B) structural isomers
 - C) diastereomers
 - D) enantiomers
 - E) None of the above

- 8. Which of the following alkyl halides would you expect to undergo S_N1 reaction most rapidly?
 - $\begin{array}{cccc} CH_3\\ CH_3CH_2CH_2CH_2-Br & CH_3CH_2CH_2CH-Br & CH_3CH_2C-Br \\ CH_3 & CH_3 & CH_3 \end{array}$ (A) (B) (C)
 - (D) They will not undergo $S_N 1$ reaction
 - (E) They react at the same rate
- 9. Which of (A)-(D) is not aromatic?



10. Which of the following is **not** aromatic?

